

RECYCLING OF COMPONENTS OF WASTED SOLID OXIDE ELECTROLYSERS OR FUEL CELLS

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Summary: Decarbonized hydrogen production is expected to play a key role in the transition to a sustainable fossil free economy ¹. However, high demand for carbon free hydrogen requires large size Solid Oxide Electrolyzers (SOEs) which in turn have economic, environmental and legislative problems of end of life products ². To solve this problem, we are attempting a strategy to recycle and reuse Solid Oxide Cells (SOCs) components and regenerate electrolytes. The separation of electrode-electrolyte assembly was performed through mechanical scratching and grinding followed by thermal and chemical treatments. Recycle materials of SOFCs components; air electrodes ($\text{La}_x\text{Sr}_{1-x}\text{CoO}_3$), nickel oxide (NiO) which constitutes about 50% in weight of the cell and Ytria Stabilized Zirconia (YSZ) were successfully obtained. Detailed crystallographic studies, microstructure, and compositions of the recovered materials were studied respectively by X-Ray Diffraction and Scanning Electron Microscope/Energy Dispersive X-Ray Spectroscopy. The electrical conductivity of the recycled electrolyte pellets was measured in air by Electrochemical Impedance Spectroscopy from 250°C to 715°C. The conductivity of the electrolyte made from recycled materials of SOC wastes was compared with different composition of commercial YSZ materials ³⁻⁵. A total electrolyte conductivity of $4.8 \times 10^{-3} \text{Scm}^{-1}$ was measured at 700°C. The lower level of conductivity of the regenerated electrolyte material is discussed and improvements to increase it are suggested.

Reference

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